

Course Outline

1. COURSE INFORMATION

Session Offered	Winter 2024	
Course Name	Chemical Engineering II: Energy Balances	
Course Code	PROC TECH 2EC3	
Date(s) and Time(s) of lectures	Monday 8:30 – 9:20, MDCL 1110 Tuesday 12:30 – 14:20, MDCL 1102	
Program Name	Automation Engineering Technology	
Calendar Description	Steady-state energy balances including possible reactions. Combined mass and energy balances. First and second law of thermodynamics. Physical/chemical equilibrium. Application to industrial processes. Drawing and interpretation of Piping and Instrumentation diagrams (P&ID).	
Instructor(s)	Lecture: Pouria Baghaei Labs: Pouria Baghaei Greg Matzke Ali Mohsenipour	Office Hours & Location: By Appointment. Emails: baghaeip@mcmaster.ca gmatzke@mcmaster.ca mohsea2@mcmaster.ca

2. COURSE SPECIFICS

Course Description			
Instruction Type	Code	Type	Hours per term
	C	Classroom instruction	39
	L	Laboratory, workshop or fieldwork	30
	T	Tutorial	
	DE	Distance education	
	Total Hours		69
Resources	ISBN	Textbook Title & Edition	Author & Publisher
	ISBN: 978-0-471-68757-3	Elementary Principles of Chemical Processes, 3rd edition, 2005 or 4th edition, 2015	<i>R. Felder and R. Rousseau, John Wiley and Sons</i>
	Other Supplies	Source	
Prerequisite(s)	<i>ENGTECH 1MC3, PROCTECH 2CE3 and registration in level II or above of Automation Engineering Technology.</i>		
Corequisite(s)	N/A		
Antirequisite(s)	N/A		
Course Specific Policies	<p>Exams: Absence from a test without an approved MSAF will result in a grade of zero for the test. If an approved MSAF is submitted, the weight of the missed test will be added to the final exam's weight. In case of multiple missed tests with approved MSAFs, the opportunity to write missed test(s) at an alternate date may be offered, at the discretion of the instructor.</p>		

	<p>Quizzes: Most quizzes will be on-line asynchronous. There might be a limited number of announced synchronous (during class time) ones. Absence from any quiz without an approved MSAF will result to a grade of zero for that quiz. A submitted MSAF for an asynchronous quiz will be accommodated by extending the submission window for that quiz. A submitted MSAF for a synchronous quiz will be accommodated either by distributing the weight of the quiz to the remaining quiz components (which might mean that a student will miss any opportunity to “drop” the worst quiz from his/her grade) or by providing the opportunity to take an equivalent asynchronous quiz.</p> <p>Lab Sessions: Labs must complete in the section/time students are registered at. Absence from a lab without an approved MSAF form will result in a grade of zero for the lab. Even if there is no lab report for this missed lab, the final lab grade will be reduced by the percentage of the total labs missed. Students may have to complete a pre-lab quiz prior to attending certain labs. Details for number of labs and lab reports will be outlined during the first week of labs.</p>
<p>Departmental Policies</p>	<p>Students must maintain a GPA of 3.5/12 to continue in the program.</p> <p>In order to achieve the required learning objectives, on average, B.Tech. students can expect to do at least 3 hours of “out-of-class” work for every scheduled hour in class. “Out-of-class” work includes reading, research, assignments and preparation for tests and examinations.</p> <p>Where group work is indicated in the course outline, such collaborative work is mandatory.</p> <p>The use of cell phones, iPods, laptops and other personal electronic devices are prohibited from the classroom during the class time, unless the instructor makes an explicit exception.</p> <p>Announcements made in class or placed on Avenue are considered to have been communicated to all students including those individuals that are not in class.</p> <p>Instructor has the right to submit work to software to identify plagiarism.</p>
<p>3. SUB TOPIC(S)</p>	
<p>Week 1</p>	<p>Review of Gas-Liquid systems with one condensable component; dew point; bubble point; equilibrium calculations</p>
<p>Week 2</p>	<p>Multi-component Gas/Liquid systems</p>
<p>Week 3</p>	<p>Solution of solids in liquids; Solubility; Crystallization</p>
<p>Week 4</p>	<p>Two liquid phases; Ternary diagram; Liquid/liquid extraction; Adsorption</p>
<p>Week 5</p>	<p>Energy Balance: Energy and first law of thermodynamics</p>
<p>Week 6</p>	<p>Open and closed system energy balance Test 1: 2 hours</p>
	<p>Mid-term Recess. No Classes</p>
<p>Week 7</p>	<p>Energy balances on non-reactive systems: Steam Tables</p>

Week 8	Energy balances on non-reactive systems: Specific Heats	
Week 9	Energy balances on non-reactive systems: Phase Change	
Week 10	Energy balances with reactions: Heat of formation and heat of reaction calculations Test 2: 2 hours	
Week 11	Energy balances with reactions: Examples	
Week 12	Combined mass and energy balances	
Week 13	Review / Overflow	

Midterm Recess: Monday, February 19 to Sunday, February 25
 Test and Examination Restriction Period: Thursday, April 4 to Thursday, April 11
 Classes end: Wednesday, April 10
 Final examination period: Friday, April 12 to Thursday, April 25
 All examinations MUST be written during the scheduled examination period

List of experiments

Lab 1	Chlorine dosing pump
Lab 2	Phosphate precipitation
Lab 3	Filter Press
Lab 4	pH adjustment
Lab 5	Distillation – PID & Instrumentation
Lab 6	Copper precipitation
Lab 7	Fluid Flow: minor losses coefficient calculation
Lab 8	Carbon Dioxide Absorption
Lab 9	Evaporator: heat economy
Lab 10	Distillation – Total Reflux

Note that this structure represents a plan and is subject to adjustment term by term. The instructor and the University reserve the right to modify elements of the course during the term. The University may change the dates and deadlines for any or all courses in extreme circumstances. If either type of modification becomes necessary, reasonable notice and communication with the students will be given with explanation and the opportunity to comment on changes.

4. ASSESSMENT OF LEARNING *including dates*	Weight
Assignments and Quizzes	10%
Mid-term tests	35%
Labs	25%
Final examination (tests cumulative knowledge)	30%
TOTAL	100%

Percentage grades will be converted to letter grades and grade points per the University calendar.

5. LEARNING OUTCOMES

1. Execute mass balance calculations on vapour/liquid, solid/liquid and liquid/liquid systems
2. Implement the multiphase equilibrium principles in calculations for liquid/liquid extraction, drying, crystallization, and other unit operations
3. Apply the concept of conservation of energy in formulation of energy balances for open and closed systems
4. Implement use of thermodynamic tables in energy balance calculations with possible phase change
5. Integrate energy balances with reaction principles and mass balances to perform calculations on processes with reactions

6. Perform preliminary design calculations for distillation columns and evaporators
7. Use instruments for performing experimental procedures and measurements.
8. Critique on the validity of theoretical predictions on different experimental settings.
9. Construct and read P&ID diagrams

6. COURSE OUTLINE – APPROVED ADVISORY STATEMENTS

ANTI-DISCRIMINATION

The Faculty of Engineering is concerned with ensuring an environment that is free of all discrimination. If there is a problem, individuals are reminded that they should contact the Department Chair, the Sexual Harassment Officer or the Human Rights Consultant, as soon as possible.

http://www.mcmaster.ca/policy/General/HR/Discrimination_Harassment_Sexual_Harassment-Prevention&Response.pdf

ACADEMIC INTEGRITY

You are expected to exhibit honesty and use ethical behaviour in all aspects of the learning process. Academic credentials you earn are rooted in principles of honesty and academic integrity. It is your responsibility to understand what constitutes academic dishonesty.

Academic dishonesty is to knowingly act or fail to act in a way that results or could result in unearned academic credit or advantage. This behaviour can result in serious consequences, e.g. the grade of zero on an assignment, loss of credit with a notation on the transcript (notation reads: “Grade of F assigned for academic dishonesty”), and/or suspension or expulsion from the university. For information on the various types of academic dishonesty please refer to the Academic Integrity Policy, located at <https://secretariat.mcmaster.ca/university-policies-procedures-guidelines/>

The following illustrates only three forms of academic dishonesty: The following illustrates only three forms of academic dishonesty:

- plagiarism, e.g. the submission of work that is not one’s own or for which other credit has been obtained.
- improper collaboration in group work.
- copying or using unauthorized aids in tests and examinations.

AUTHENTICITY / PLAGIARISM DETECTION

Some courses may use a web-based service (Turnitin.com) to reveal authenticity and ownership of student submitted work. For courses using such software, students will be expected to submit their work electronically either directly to Turnitin.com or via an online learning platform (e.g. A2L, etc.) using plagiarism detection (a service supported by Turnitin.com) so it can be checked for academic dishonesty.

Students who do not wish their work to be submitted through the plagiarism detection software must inform the Instructor before the assignment is due. No penalty will be assigned to a student who does not submit work to the plagiarism detection software. All submitted work is subject to normal verification that standards of academic integrity have been upheld (e.g., on-line search, other software, etc.). For more details about McMaster’s use of Turnitin.com, please go to www.mcmaster.ca/academicintegrity.

COURSES WITH AN ON-LINE ELEMENT

Some courses may use on-line elements (e.g. e-mail, Avenue to Learn (A2L), LearnLink, web pages, capa, Moodle, ThinkingCap, etc.). Students should be aware that, when they access the electronic components of a course using these elements, private information such as first and last names, user names for the McMaster e-mail accounts, and program affiliation may become apparent to all other students in the same course. The available information is dependent on the technology used. Continuation in a course that uses on-line elements will be deemed consent to this disclosure. If you have any questions or concerns about such disclosure please discuss this with the course instructor.

ONLINE PROCTORING

Some courses may use online proctoring software for tests and exams. This software may require students to turn on their video camera, present identification, monitor and record their computer activities, and/or lock/restrict their browser or other applications/software during tests or exams. This software may be required to be installed before the test/exam begins.

COMMUNICATIONS

It is the student's responsibility to:

- Maintain current contact information with the University, including address, phone numbers, and emergency contact information.
- Use the University provided e-mail address or maintain a valid forwarding e-mail address.
- Regularly check the official University communications channels. Official University communications are considered received if sent by postal mail, by fax, or by e-mail to the student's designated primary e-mail account via their @mcmaster.ca alias.
- Accept that forwarded e-mails may be lost and that e-mail is considered received if sent via the student's @mcmaster.ca alias.
- Check the McMaster/Avenue email and course websites on a regular basis during the term.

CONDUCT EXPECTATIONS

As a McMaster student, you have the right to experience, and the responsibility to demonstrate, respectful and dignified interactions within all of our living, learning and working communities. These expectations are described in the Code of Student Rights & Responsibilities (the "Code"). All students share the responsibility of maintaining a positive environment for the academic and personal growth of all McMaster community members, whether in person or online.

It is essential that students be mindful of their interactions online, as the Code remains in effect in virtual learning environments. The Code applies to any interactions that adversely affect, disrupt, or interfere with reasonable participation in University activities. Student disruptions or behaviours that interfere with university functions on online platforms (e.g. use of Avenue 2 Learn, WebEx or Zoom for delivery), will be taken very seriously and will be investigated. Outcomes may include restriction or removal of the involved students' access to these platforms.

ACADEMIC ACCOMMODATION OF STUDENTS WITH DISABILITIES

Students with disabilities who require academic accommodation must contact Student Accessibility Services (SAS) at 905-525-9140 ext. 28652 or sas@mcmaster.ca to make arrangements with a Program Coordinator. For further information, consult McMaster University's Academic Accommodation of Students with Disabilities policy.

REQUESTS FOR RELIEF FOR MISSED ACADEMIC TERM WORK

McMaster Student Absence Form (MSAF): In the event of an absence for medical or other reasons, students should review and follow the Academic Regulation in the Undergraduate Calendar "Requests for Relief for Missed Academic Term Work".

ACADEMIC ACCOMMODATION FOR RELIGIOUS, INDIGENOUS OR SPIRITUAL OBSERVANCES (RISO)

Students requiring academic accommodation based on religious, indigenous or spiritual observances should follow the procedures set out in the RISO policy. Students should submit their request to their Faculty Office normally within 10 working days of the beginning of term in which they anticipate a need for accommodation or to the Registrar's Office prior to their examinations. Students should also contact their instructors as soon as possible to make alternative arrangements for classes, assignments, and tests. <http://www.mcmaster.ca/policy/Students-AcademicStudies/Studentcode.pdf>

COPYRIGHT AND RECORDING

Students are advised that lectures, demonstrations, performances, and any other course material provided by an instructor include copyright protected works. The Copyright Act and copyright law protect every original literary, dramatic, musical and artistic work, including lectures by University instructors

The recording of lectures, tutorials, or other methods of instruction may occur during a course. Recording may be done by either the instructor for the purpose of authorized distribution, or by a student for the purpose of personal study. Students should be aware that their voice and/or image may be recorded by others during the class. Please speak with the instructor if this is a concern for you.

EXTREME CIRCUMSTANCES

The University reserves the right to change the dates and deadlines for any or all courses in extreme circumstances (e.g., severe weather, labour disruptions, etc.). Changes will be communicated through regular McMaster communication channels, such as McMaster Daily News, A2L and/or McMaster email.