CALENDAR/COURSE DESCRIPTION
An introduction to statistical distributions and their properties, and the statistical basis of thermodynamics at the microscopic level, with applications to problems originating in a modern laboratory or engineering environment.

PRE-REQUISITES AND ANTI-REQUISITES
Prerequisite(s): Credit or registration in ENGPHYS 2NE3, 2QM3, and 3L04
Antirequisite(s): ENGPHYS 2H04, PHYSICS 2H04, PHYSICS 3K03

INSTRUCTOR OFFICE HOURS AND CONTACT INFORMATION
Dr. Ryan Lewis
JHE A323
rlewis@mcmaster.ca
ext. 24923
Office Hours:
By appointment

TEACHING ASSISTANT OFFICE HOURS AND CONTACT INFORMATION
Luis Abalo Sangervasi
abalosal@mcmaster.ca
Office Hours:
See course website
Lai Xu
xul71@mcmaster.ca
See course website

COURSE WEBSITE/ALTERNATE METHODS OF COMMUNICATION
http://avenue.mcmaster.ca/

COURSE INTENDED LEARNING OUTCOMES
By the end of this course, students should be able to:
- Explain the statistical underpinnings of thermodynamics
- Apply combinatorics to determine the microstates and macrostates of a system
- Explain the origin of equipartition theorem and use it to predict properties of a system
- Apply Boltzmann and quantum statistics to a wide range of topics (e.g., heat capacity, electrons in semiconductors, magnetism, blackbody radiation, adsorption, kinetic theory of gases)

MATERIALS AND FEES
Required Texts:
“An Introduction to Thermal Physics” by Daniel V. Schroeder
Calculator:
Only the McMaster Standard Calculator will be permitted in tests and examinations. This is available at the Campus Store.

**Course Format and Expectations**

The course is organized as follows:
- 3 lectures per week
- 1 tutorial per week
- 6 assignments
- 1 midterm test
- Final exam

**Course Schedule**

<table>
<thead>
<tr>
<th>Week</th>
<th>Readings</th>
<th>Tutorial</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>The first law (ideal gas, thermal equilibrium, heat capacity) Schroeder ch. 1 No</td>
<td></td>
</tr>
<tr>
<td>2</td>
<td>Entropy and the second law (combinatorics, Einstein solid, large systems) Schroeder ch. 2 Yes</td>
<td></td>
</tr>
<tr>
<td>3</td>
<td>Statistical model of ideal gas, entropy, temperature Schroeder ch. 2.3 Yes</td>
<td></td>
</tr>
<tr>
<td>4</td>
<td>Paramagnetism, mechanical equilibrium, chemical equilibrium Schroeder ch. 3 Yes</td>
<td></td>
</tr>
<tr>
<td>5</td>
<td>Free energy and chemical thermodynamics (chemical potential, free energy, phase transformations) Schroeder ch. 4 Yes</td>
<td></td>
</tr>
<tr>
<td>6</td>
<td>Boltzmann statistics (Boltzmann factor, equipartition and the Maxwell speed distribution) Schroeder ch. 6 Yes</td>
<td></td>
</tr>
<tr>
<td>7</td>
<td>Winter break</td>
<td></td>
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<tr>
<td>8</td>
<td>Boltzmann statistics (Free energy, ideal gas, rates) Schroeder ch. 6 Yes</td>
<td></td>
</tr>
<tr>
<td>9</td>
<td>Quantum statistics (Gibbs factor, Bosons and Fermions, photons) Schroeder ch. 7 Yes</td>
<td></td>
</tr>
<tr>
<td>10</td>
<td>Quantum statistics (Blackbody radiation, degenerate Fermi gases) Schroeder ch. 7 Yes</td>
<td></td>
</tr>
<tr>
<td>11</td>
<td>Free electron model, Bloch’s theorem</td>
<td></td>
</tr>
<tr>
<td>12</td>
<td>Electronic band structure</td>
<td>Yes</td>
</tr>
<tr>
<td>13</td>
<td>Bose–Einstein condensation Schroeder ch. 7 Yes</td>
<td></td>
</tr>
<tr>
<td>14</td>
<td>Review TBD</td>
<td></td>
</tr>
</tbody>
</table>

**Assessment**

<table>
<thead>
<tr>
<th>Component</th>
<th>Weight</th>
</tr>
</thead>
<tbody>
<tr>
<td>Participation</td>
<td>5%</td>
</tr>
<tr>
<td>Assignments</td>
<td>30%</td>
</tr>
<tr>
<td>Midterm test</td>
<td>25%</td>
</tr>
<tr>
<td>Final exam</td>
<td>40%</td>
</tr>
<tr>
<td>Total</td>
<td>100%</td>
</tr>
</tbody>
</table>
You are encouraged to discuss assignments with your classmates. However, posting and sharing questions/answers on the internet (e.g., chegg.com and similar sites) for is strictly forbidden. In addition, the use of generative AI (e.g., ChatGPT) to complete assignments is not allowed. The calculations and the assignment that you hand in must be your own work. In carrying out calculations on tests or assignments, it is important to explain what you are doing. In addition to helping with learning, explaining a rational thought process will score points even if the calculation is incorrect.

**ACCREDITATION LEARNING OUTCOMES**

The Learning Outcomes defined in this section are measured for Accreditation purposes only and will not be directly taken into consideration in determining a student’s grade in the course.

<table>
<thead>
<tr>
<th>Outcomes</th>
<th>Indicators</th>
</tr>
</thead>
<tbody>
<tr>
<td>Explain the laws of thermodynamics and outline their statistical underpinnings, discriminate between basic models in statistical mechanics (e.g., Einstein solid, Debye solid)</td>
<td>1.2</td>
</tr>
<tr>
<td>Apply Boltzmann and quantum statistics to calculate properties of a system</td>
<td>1.4</td>
</tr>
<tr>
<td>Identify appropriate statistical mechanics tools suitable to apply to a range of physical systems</td>
<td>2.1</td>
</tr>
</tbody>
</table>

For more information on Accreditation, please visit: [https://www.engineerscanada.ca](https://www.engineerscanada.ca)

**EQUITY, DIVERSITY, AND INCLUSION**

Every registered student belongs in this course. Diversity of backgrounds and experiences is expected and welcome. You can expect your Instructor to be respectful of this diversity in all aspects of the course, and the same is expected of you.

The Department of Engineering Physics is committed to creating an environment in which students of all genders, cultures, ethnicities, races, sexual orientations, abilities, and socioeconomic backgrounds have equal access to education and are welcomed and treated fairly. If you have any concerns regarding inclusion in our Department, in particular if you or one of your peers is experiencing harassment or discrimination, you are encouraged to contact the Chair, Associate Undergraduate Chair, Academic Advisor or to contact the [Equity and Inclusion Office](https://www.engineerscanada.ca).

**PHYSICAL AND MENTAL HEALTH**

For a list of McMaster University’s resources, please refer to the [Student Wellness Centre](https://www.engineerscanada.ca).

**ACADEMIC INTEGRITY**

You are expected to exhibit honesty and use ethical behaviour in all aspects of the learning process. Academic credentials you earn are rooted in principles of honesty and academic integrity. **It is your responsibility to understand what constitutes academic dishonesty.**

Academic dishonesty is to knowingly act or fail to act in a way that results or could result in unearned academic credit or advantage. This behaviour can result in serious consequences, e.g. the grade of zero on an assignment, loss of credit with a notation on the transcript (notation reads: “Grade of F assigned for academic dishonesty”), and/or suspension or expulsion from the university. For information on the various types of academic dishonesty please refer to the [Academic Integrity Policy](https://www.engineerscanada.ca), located at [https://secretariat.mcmaster.ca/university-policies-procedures-guidelines/](https://secretariat.mcmaster.ca/university-policies-procedures-guidelines/).
The following illustrates only three forms of academic dishonesty:
1. plagiarism, e.g. the submission of work that is not one’s own or for which other credit has been obtained.
2. improper collaboration in group work.
3. copying or using unauthorized aids in tests and examinations.

**COURSES WITH AN ON-LINE ELEMENT**

McMaster is committed to an inclusive and respectful community. These principles and expectations extend to online activities including electronic chat groups, video calls and other learning platforms.

Some courses may use on-line elements (e.g. e-mail, Avenue to Learn (A2L), LearnLink, web pages, capa, Moodle, ThinkingCap, etc.). Students should be aware that, when they access the electronic components of a course using these elements, private information such as first and last names, user names for the McMaster e-mail accounts, and program affiliation may become apparent to all other students in the same course. The available information is dependent on the technology used. Continuation in a course that uses on-line elements will be deemed consent to this disclosure. If you have any questions or concerns about such disclosure, please discuss this with the course instructor.

**ONLINE PROCTORING**

Some courses may use online proctoring software for tests and exams. This software may require students to turn on their video camera, present identification, monitor and record their computer activities, and/or lock/restrict their browser or other applications/software during tests or exams. This software may be required to be installed before the test/exam begins.

**CONDUCT EXPECTATIONS**

As a McMaster student, you have the right to experience, and the responsibility to demonstrate, respectful and dignified interactions within all of our living, learning and working communities. These expectations are described in the Code of Student Rights & Responsibilities (the “Code”). All students share the responsibility of maintaining a positive environment for the academic and personal growth of all McMaster community members, whether in person or online.

It is essential that students be mindful of their interactions online, as the Code remains in effect in virtual learning environments. The Code applies to any interactions that adversely affect, disrupt, or interfere with reasonable participation in University activities. Student disruptions or behaviours that interfere with university functions on online platforms (e.g. use of Avenue 2 Learn, WebEx or Zoom for delivery), will be taken very seriously and will be investigated. Outcomes may include restriction or removal of the involved students’ access to these platforms.

**ACADEMIC ACCOMMODATION OF STUDENTS WITH DISABILITIES**

Students with disabilities who require academic accommodation must contact Student Accessibility Services (SAS) at 905-525-9140 ext. 28652 or sas@mcmaster.ca to make arrangements with a Program Coordinator. For further information, consult McMaster University’s Academic Accommodation of Students with Disabilities policy.

**COURSE POLICY ON MISSED WORK, EXTENSIONS, AND LATE PENALTIES**
1. It is the students’ responsibility to regularly check the course webpage (Avenue to Learn) for updates and announcements.
2. Work handed in 0–24 h after a deadline will receive a 25% late penalty. Work submitted > 24 h after a deadline will receive an automatic mark of 0.

**SUBMISSION OF REQUEST FOR RELIEF FOR MISSED ACADEMIC WORK**

In the event of an absence for medical or other reasons, students should review and follow the Academic Regulation in the Undergraduate Calendar “Requests for Relief for Missed Academic Term Work”.

1. Relief for missed academic work worth less than 25% of the final grade resulting from medical or personal situations lasting up to three calendar days:
   - Use the McMaster Student Absence Form (MSAF) on-line self-reporting tool. No further documentation is required.
   - Students may submit requests for relief using the MSAF once per term.
   - An automated email will be sent to the course instructor, who will determine the appropriate relief. Students must immediately follow up with their instructors. Failure to do so may negate the opportunity for relief.
   - The MSAF cannot be used to meet a religious obligation or to celebrate an important religious holiday.
   - The MSAF cannot be used for academic work that has already been completed attempted.
   - An MSAF applies only to work that is due within the period for which the MSAF applies, i.e. the 3-day period that is specified in the MSAF; however, all work due in that period can be covered by one MSAF.
   - The MSAF cannot be used to apply for relief for any final examination or its equivalent. See Petitions for Special Consideration above.

2. For medical or personal situations lasting more than three calendar days, and/or for missed academic work worth 25% or more of the final grade, and/or for any request for relief in a term where the MSAF has been used previously in that term:
   - Students must report to their Faculty Office to discuss their situation and will be required to provide appropriate supporting documentation.
   - If warranted, the Faculty Office will approve the absence, and the instructor will determine appropriate relief.

**ACADEMIC ACCOMMODATION FOR RELIGIOUS, INDIGENOUS OR SPIRITUAL OBSERVANCES (RISO)**

Students requiring academic accommodation based on religious, indigenous or spiritual observances should follow the procedures set out in the RISO policy. Students should submit their request to their Faculty Office normally within 10 working days of the beginning of term in which they anticipate a need for accommodation or to the Registrar’s Office prior to their examinations. Students should also contact their instructors as soon as possible to make alternative arrangements for classes, assignments, and tests.

**COPYRIGHT AND RECORDING**

Students are advised that lectures, demonstrations, performances, and any other course material provided by an instructor include copyright protected works. The Copyright Act and copyright law protect every original literary, dramatic, musical and artistic work, including lectures by University instructors.

The recording of lectures, tutorials, or other methods of instruction may occur during a course. Recording may be done by either the instructor for the purpose of authorized distribution, or by a student for the purpose of personal
study. Students should be aware that their voice and/or image may be recorded by others during the class. Please speak with the instructor if this is a concern for you.

**EXTREME CIRCUMSTANCES**

The University reserves the right to change the dates and deadlines for any or all courses in extreme circumstances (e.g., severe weather, labour disruptions, etc.). Changes will be communicated through regular McMaster communication channels, such as McMaster Daily News, A2L and/or McMaster email.